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Up-conversion Luminescence of Terbium(III) in SiO₂ and ZnO–SiO₂ Glasses Induced by Simultaneous Absorption of Three Photons and Saturation

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ABSTRACT Green up-conversion luminescence (UCL) of Tb³⁺ in SiO₂ and ZnO–SiO₂ glasses was observed with femtosecond (fs) laser (800 nm) excitation. The UCL mechanism of Tb³⁺ in both glasses contributed to the simultaneous three-photon absorption. The saturation effect of Tb³⁺ in SiO₂ glass took place. The saturation effect of excited-state levels of Tb³⁺ in SiO₂ glass was associated with the slower decay rate of ⁵D₄ level. This result implied that saturation power of Tb³⁺ in ZnO–SiO₂ glass was higher than that in SiO₂ glass.

KEYWORDS saturation, Terbium, up-conversion

The key feature of UCL of rare earth (RE) ions is their ability to absorb and combine two or more photons of low near-infrared (NIR) energy to emit a higher energy visible photon. UCL of RE ions in transparent hosts has been intensively investigated because it has many potential applications in up-conversion laser, displays, fluorescent probe, optical communications and fiber-optic amplifiers, under forth.^[1,2] In the past decades, research into UCL was mainly focused on the RE ions with plenty of levels, such as Er³⁺, Ho³⁺, Tm³⁺, and Nd³⁺ ions.^[3–5] Reports on UCL of Tb³⁺ were very rare because the energy levels of Tb³⁺ ions cannot easily match those of other RE ions. In previous work, two-photon UCL of Tb³⁺ through energy transfer from Yb³⁺ or ligand to Tb³⁺ at NIR light excitation has been reported.^[6–11] But, to our knowledge, virtually nothing has been reported on UCL of Tb³⁺ through three-photon excitation by fs laser. Because Tb³⁺ ion-doped hosts are universally applied to commercial green phosphors, the research on UCL is partly motivated by the technological drive toward novel phosphors and partly because the scientific community would like to understand the fundamental physics. Because ultrafast titanium–sapphire tunable NIR laser can produce fs pulse and provide high density to be used as excitation resource, it is an ideal tool to study the optical

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properties of materials.^[12–14] You et al. reported the UCL of Eu^{3+} and Ce^{3+} in sol-gel Al_2O_3 – SiO_2 glasses due to three-photon simultaneous absorption.^[15,16] In this paper, we report the UCL of Tb^{3+} ions in SiO_2 and ZnO – SiO_2 glasses at fs laser excitation and compare their UCL characteristics. It is important that we observed higher saturation power of Tb^{3+} in ZnO – SiO_2 glass than its SiO_2 glass. It can enrich the understanding of UCL processes of Tb^{3+} .

SiO_2 and 10ZnO–90 SiO_2 glasses containing 1% Tb_2O_3 (2.49 mol/m^3) were prepared by a sol-gel method. Tetraethoxysilane (TEOS), $\text{ZnCl}_2 \cdot 6\text{H}_2\text{O}$, and $\text{TbCl}_3 \cdot 6\text{H}_2\text{O}$ were used as starting materials. First, the solution containing TEOS, ethanol, and deionized water (containing 0.15 mol/L HCl) was vigorously stirred for 1 h at room temperature. $\text{ZnCl}_2 \cdot 6\text{H}_2\text{O}$ and $\text{TbCl}_3 \cdot 6\text{H}_2\text{O}$ dissolved in ethanol were added to the above solution and stirred for one-half hour. Then, the mixed solution of H_2O and ethanol with the mole ratio of 4:1 was added, and the resultant solution was stirred for another one-half hour to obtain homogeneous solution. The (obtained above solution) was cast into a plastic container and placed at room temperature for about 4 weeks to form a stiff gel of 1–3 mm thickness. In order to completely hydrolyze the alkoxide, the gel was heated to 150°C in a sealed vessel together with water and held for 30 h. Finally, the gel was sintered at 600°C for 5 h.

The excitation spectra were recorded using monochromator (Jobin Yvon, HR 320, Japan) and a photomultiplier (Hamamatsu, R955). A 500 W xenon lamp with light passed through a monochromator (H20, Jobin Yvon) was used for excitation source. UCL spectra and dynamics processes were measured by fs pulse from a Ti:sapphire regenerative amplifier laser system (Hurricane, Spectra Physics) operating at a wavelength of 800 nm with a 1-kHz repetition rate and approximately 130-fs pulse duration. The power density can be varied from 0.15 to 0.8 W/mm^2 .

Figure 1 shows the UCL spectra of SiO_2 and ZnO – SiO_2 containing Tb_2O_3 at femtosecond laser excitation. It is clear that no emission in ZnO – SiO_2 glass with the absence of Tb^{3+} was observed, and strong green emissions were observed in SiO_2 and ZnO – SiO_2 glasses containing Tb^{3+} . Four group lines at 488, 544, 583, and 620 nm were observed, which were from $^5\text{D}_4$ – $^7\text{F}_j$ ($j = 6, 5, 4, 3$) transition emissions.

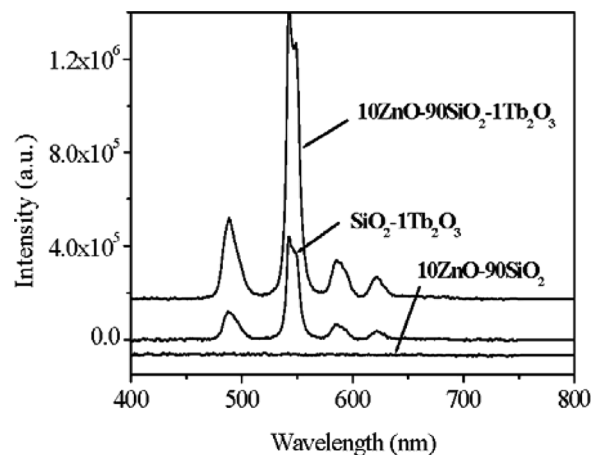


FIGURE 1 UCL spectra of Tb^{3+} in SiO_2 and ZnO – SiO_2 glasses at femtosecond laser excitation. Excitation wavelength is 800 nm and power is 200 mW/mm^2 .

The strongest emission locating at 544 nm was from $^5\text{D}_4$ – $^7\text{F}_5$ transitions.

To better understand and compare the mechanism of UCL of Tb^{3+} in SiO_2 and ZnO – SiO_2 glasses, the UCL integrated intensity of $^5\text{D}_4$ – $^7\text{F}_j$ transition emissions as a function of the excitation power was studied. The UCL intensity can be expressed as follows,

$$I \propto p^n$$

where I is the integrated intensity of UCL, p is the pump power of fs laser, and n is the number of photons absorbed per visible photon emitted for any UCL mechanism. According to the above equation, n is the slope of the log-log plot between the UCL intensity and pump power. Figure 2 is the UCL intensity as a function of pump power. It can

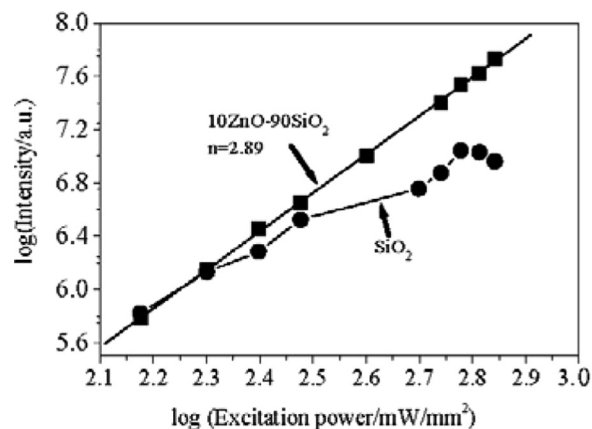


FIGURE 2 Dependence of UCL intensity on the pump power.

be seen that the three-photon excited luminescence of Tb^{3+} in SiO_2 glass saturates at high excitation powers. The slope of the logarithmic line is determined by linear fitting to be about 2.89 in ZnO-SiO_2 glass. This result indicated that the three-photon excitation in ZnO-SiO_2 glass predominant.

To clarify the mechanism of UCL processes, the excited-state energy levels should be known. Figure 3 shows the (a) absorption and (b) excitation spectra of Tb^{3+} in glasses monitoring 544 nm emission. In excitation spectra, the broad band ranging from 220 to 290 nm centered at 260 nm was associated with $4f-5d$ transitions of Tb^{3+} ions. There existed many weak excitation lines of Tb^{3+} ions in ultraviolet range (300–380 nm), which were associated with $^7\text{F}_6-^5\text{D}_3$, $^5\text{G}_J$, $^5\text{L}_6$ transitions. In absorption spectra (Fig. 3(a)), only $4f-5d$ excitation band (200–300 nm) and weak absorption lines (300–380 nm) were observed.

In ZnO-SiO_2 glass, three-photon excitation predominated. To our knowledge, this is the first report on three-photon UCL of Tb^{3+} in glasses or other hosts. For three-photon UCL processes, there existed two possible mechanisms.^[17] One is the excited-state absorption (ESA). For ESA mechanisms, the material must have absorption at about 800 nm and 400 nm, respectively. The second is the simultaneous three-photon absorption. Because there is no absorption at 400 nm and 800 nm, thus we can exclude the ESA mechanism. We should further consider the mechanism of the simultaneous three-photon absorption. The simultaneous three-photon absorption

requires an excited state that matches the three-photon energy, and high photon density. In previous references, this mechanism attracted little attention.^[18] According to excitation spectra, the simultaneous three-photon absorption corresponded with resonant $4f-5d$ transitions absorption (about 266 nm) of Tb^{3+} at fs laser (800 nm) excitation. In addition, fs laser can provide ultrafast pulse and high-density laser. Thus we suggested that the UCL mechanism of Tb^{3+} in ZnO-SiO_2 was three-photon simultaneous absorption.

Detailed UCL processes in ZnO-SiO_2 glass are shown in Fig. 4. First, the electrons at the ground state $^7\text{F}_6$ level were resonantly excited onto upper $5d$ excited state through the simultaneous three-photon absorption. Then electrons at $5d$ excited state may nonradiatively relax fast $^5\text{D}_1$, $^5\text{D}_3$, and $^5\text{D}_4$ level, and as a consequence $^5\text{D}_4-^7\text{F}_J$ transition emissions follow.

In SiO_2 glass, no absorption at 800 nm and 400 nm was observed. Two-photon excitation cannot take place, and three-photon excitation resonantly matched $4f-5d$ absorption. We suggest that UCL mechanism of Tb^{3+} in SiO_2 glass was also simultaneous three-photon absorption. Detailed UCL process was similar with that in ZnO-SiO_2 glasses. According to Fig. 2, in SiO_2 glass, saturation effect occurred. This contributed to the competition between the decay of $^5\text{D}_4$ level and the feed from higher excited state, leading to the saturation of $^5\text{D}_4$ level. Due to ultrafast pulse and high density of

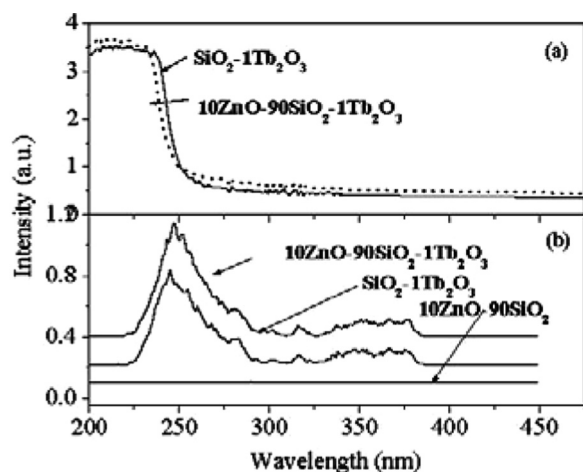


FIGURE 3 The (a) absorption and (b) excitation spectra of Tb^{3+} monitoring 544 nm emission.

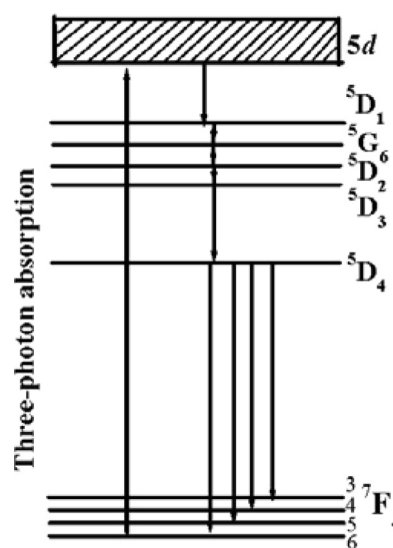


FIGURE 4 The level diagram of Tb^{3+} and UCL process.

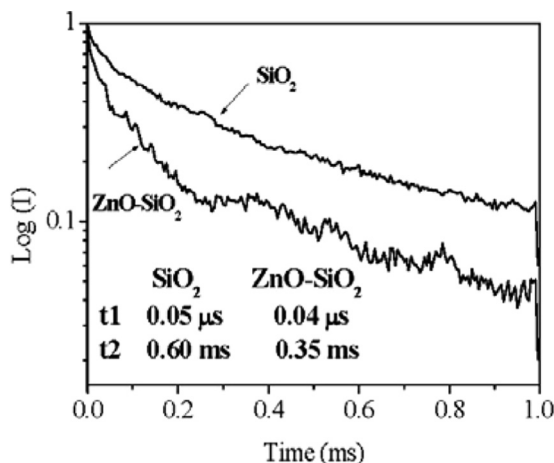


FIGURE 5 Decay curves of Tb³⁺ monitoring 544 nm at femto-second laser excitation.

fs laser, the excitation of three-simultaneous absorption and nonradiative relaxation from $5d$ excited state to 5D_3 , followed to 5D_4 is fast. The intensity of $^5D_4 \rightarrow ^7F_J$ on pump power mainly depended on the decay of 5D_4 level. Figure 5 shows the decay curves of Tb³⁺ monitoring 544 nm. It can be seen that 5D_4 level second-exponentially decayed. The lifetimes were obtained by second-exponential fitting and are listed in Fig. 5. The fast process was attributed to the nonradiative relaxation from 5D_4 to some quenching centers. The slow process corresponded with $^5D_4 \rightarrow ^7F_J$ transitions. The lifetime of fast and slow process of Tb³⁺ in SiO₂ glass was longer than that in ZnO-SiO₂ glass, indicating slower decay rate in SiO₂ glass, leading to the observation that the saturation of 5D_4 level of Tb³⁺ in SiO₂ glass is easier to occur than is that in ZnO-SiO₂ glass. Thus the UCL was the dominant depletion mechanism of 5D_4 in SiO₂ glass, leading to saturation effect to occur. This saturation effect of UCL was also observed in Er³⁺-Yb³⁺ codoped Gd₂O₃ nanocrystals.^[19] It implied that higher saturation power of Tb³⁺ is obtained in ZnO-SiO₂ glass in comparison with SiO₂ glass.

In conclusion, SiO₂ and ZnO-SiO₂ glasses containing terbium ions were prepared by a sol-gel method, and their UCL of Tb³⁺ ions was analyzed and compared at fs laser excitation. In both glasses, the intensive green UCL of Tb³⁺ was observed. The results of dependence of UCL intensity on pump power indicated that the three-photon UCL processes in ZnO-SiO₂ glass predominated and saturation effect took place in SiO₂ glass. In view of the energy level of

Tb³⁺, the mechanism of UCL of Tb³⁺ in both glasses is simultaneous three-photon excitation. However, the saturation of 5D_4 population of Tb³⁺ ions in SiO₂ glass occurred as a result of their slow decay rate in this glass, leading to a marked deviation from the cubic dependence of the emission intensity on the excitation power.

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